

Combinatorial synthesis of anilinoanthraquinone derivatives and evaluation as non-nucleotide-derived P2Y₂ receptor antagonists

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Abstract—A library of anilinoanthraquinone derivatives was synthesized by parallel Ullmann coupling reaction of bromaminic acid with aniline derivatives in solution using a compact parallel synthesizer. The products were purified by HPLC and evaluated as antagonists at mouse and human P2Y₂ receptors. 4-Phenylamino-substituted 1-amino-2-sulfoanthraquinones, for example, 1-amino-4-(2-methoxyphenyl)-2-sulfoanthraquinone (PSB-716), were potent P2Y₂ antagonists with IC₅₀ values in the low micromolar range.

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P2Y₂ receptors (P2Y₂R) belong to the family of G protein-coupled nucleotide (P2) receptors.^{1–4} They are activated by the physiological nucleotides UTP and ATP, and by dinucleotides, such as diadenosine tetraphosphate (Ap₄A).⁴ P2Y₂R show a wide distribution in the body, including lung, heart, skeletal muscle, spleen, kidney, and brain.^{1–4} There is a lack of potent and selective P2Y₂R antagonists, which are required as pharmacological tools to elucidate the (patho)physiological roles of the receptors.^{1–5} In addition, such compounds have potential as novel therapeutics, for example, as anti-inflammatory agents, for the treatment of coronary vasospastic disorders, as analgesics, or as neuroprotective drugs.^{1–6}

The present study focuses on the development of non-nucleotide-derived P2Y₂R antagonists using Reactive Blue 2 (RB-2, **1**, Fig. 1) as a lead structure. RB-2 is one of the most potent P2Y₂ antagonists known to date (IC₅₀ 1–5 μM).⁴ However, RB-2 also inhibits ectonucleotidases⁷ and blocks other P2 receptor subtypes as well.^{3,4,8,9} Furthermore, RB-2 has a relatively high molecular weight (MW = 840 g/mol) and bears three

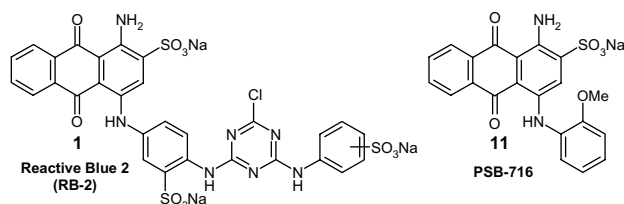


Figure 1. Structures of P2Y₂ receptor antagonists Reactive Blue 2 (RB-2) and compound **11** (PSB-716).

negatively charged sulfonate groups. Therefore, it does not exhibit properties that are desirable for a drug (MW < 500 g/mol, not permanently charged at pH 7.4).¹⁰

In previous studies anthraquinone derivatives related to RB-2, including Acid Blue 25 (AB-25, compound **3**, Table 2), have been evaluated as P2 receptor antagonists and ectonucleotidase inhibitors.^{9,11} The substitution pattern on the aniline ring has been found to be important for activity and P2Y versus P2X receptor selectivity.^{11b–d} However, no structure–activity relationships (SARs) have been reported for this class of compounds at P2Y₂R.

In the present study, we investigated whether a combinatorial synthetic approach using a compact parallel

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Table 1. Yields of anthraquinone derivatives synthesized in a compact parallel synthesizer in solution

Compound	R	Yield ^a
3		60
4		74
5		83
6		26
7		72
8		42
9		26
10		76
11 (PSB-716)		67
12		82
13		52
15		61
15		16
16		56

Table 1 (continued)

Compound	R	Yield ^a
17		35
18		74
19		17
20		90

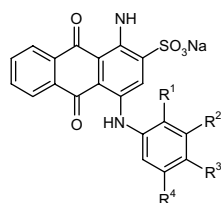
^a Isolated yields (purity of products $\geq 98\%$) calculated based on starting compound **2**; true yields were higher since commercial **2** was only 90% pure (main contaminant: desbromo derivative, 8%).¹⁵

synthesizer for the preparation of a library of 4-phenylamino-substituted 1-amino-2-sulfoanthraquinones would be feasible. Furthermore, we developed an efficient HPLC method for the purification of the products. The compounds were evaluated as P2Y₂R antagonists and SARs were analyzed.

The classical method to obtain the target compounds is the copper-catalyzed Ullmann coupling reaction.¹² Thus, bromaminic acid sodium salt (**2**) was reacted with aniline derivatives in the presence of copper(II) sulfate and sodium carbonate in water at 90 °C for 2 days. These conditions were found to be a good compromise for all the aniline derivatives employed, which possess very different reactivities. Parallel synthesis was performed in polypropylene vials using a MiniBlockTM synthesizer (Mettler Toledo, Switzerland).¹³ After lyophilization the products were subjected to purification by HPLC on a Eurospher 100 C 18 column (10 μ m, 250 \times 20 mm).¹⁴

Examples for typical chromatograms are shown in [Figure 2](#) for products **7**, **8**, and **13**. The starting compound bromaminic acid (**2**, red) and the side-product 1-amino-4-hydroxy-2-sulfoanthraquinone (**21**, dark red), which results from a substitution of the bromine atom by hydroxide, eluted within the first 5 min. In contrast, the somewhat less polar anilino derivatives **3–20** were eluted between 8 and 14 min thus allowing a baseline separation from the starting compound and the side-product ([Fig. 2](#)), which results in very pure products. Purities were determined by LCMS (for details, see [Supplementary Information](#)) and found to be $>98\%$ in all cases. The products were further characterized by ¹H- and ¹³C NMR spectra, elemental analysis, or high resolution mass spectra, respectively; for a typical example, see footnote.¹⁶

As shown in [Table 1](#) yields ranged from 16 to 90% depending on the substituents on the phenyl ring. In most cases, yields were greater than 50%; only for 6 products (**6**, **8**, **9**, **15**, **17**, and **19**) lower yields were ob-

Table 2. Pharmacological Evaluation of the synthesized anthraquinone derivatives as antagonists at mouse and human P2Y₂ receptors

Compound	R ¹	R ²	R ³	R ⁴	IC ₅₀ ± SEM (mP2Y ₂) ^a [μM]	IC ₅₀ ± SEM (hP2Y ₂) ^b [μM]
1 (RB-2)		For structure, see Figure 1			5.0 ± 0.5	1.85 ± 0.39
2^c		For structure, see Table 1			>100	nd ^d
3 (AB-25)	H	H	H	H	11.1 ± 3.0	5.61 ± 0.47
4	CH ₃	H	H	H	10 ± 2	16.2 ± 2.2
5	H	CH ₃	H	H	22 ± 7	3.04 ± 0.58
6	H	H	CH ₃	H	27 ± 5	24.9 ± 8.19
7	CH ₃	CH ₃	H	H	14 ± 9	7.95 ± 1.05
8	CH ₃	H	CH ₃	H	9 ± 1	5.43 ± 1.78
9	CH ₃	H	H	CH ₃	17 ± 5	30.3 ± 9.79
10	C ₂ H ₅	H	H	H	24 ± 4	9.26 ± 2.76
11 (PSB-716)	OCH ₃	H	H	H	9 ± 2	9.82 ± 0.43
12	H	H	OCH ₃	H	17 ± 1	11.5 ± 2.43
13	OC ₂ H ₅	H	H	H	17 ± 4	15.0 ± 5.6
14	H	H	OC ₂ H ₅	H	9 ± 1	11.5 ± 2.43
15	H	Cl	H	H	>100	>30
16	H	H	Cl	H	20 ± 7	5.31 ± 0.54
17	H	NO ₂	H	H	>100	7.52 ± 0.82
18	H	H	COCH ₃	H	>100	nd ^d
19	H	H	NHC ₆ H ₅	H	30 ± 7	3.93 ± 0.76
20	H	H	NHCOCH ₃	H	12 ± 4	5.81 ± 1.57

^a Inhibition of Ca²⁺ mobilization induced by 3 μM UTP in NG108-15 cells which natively express the mouse P2Y₂ receptor.

^b Inhibition of Ca²⁺ mobilization induced by 1 μM UTP in 1321N1 astrocytoma cells recombinantly expressing the human P2Y₂ receptor. IC₅₀ values for antagonists were calculated by nonlinear regression using Prism[®] 3.0.

^c Bromaminic acid (see Table 1).

^d nd, not determined.

tained. In case of compounds **15** and **17** electronic reasons (electron-withdrawing substituents) are likely, while steric hindrance might be an explanation for the low yields of **8** and **9**; however other ortho-substituted anilines provided higher yields. Nevertheless, all desired products were obtained in sufficient amount and purity for pharmacological testing.

P2Y₂R are coupled to phospholipase Cβ₁ via Gα_{q/11} protein mediating the production of inositol trisphosphate which in turn leads to intracellular calcium release.³ All products were evaluated as antagonists at mouse P2Y₂R natively expressed in neuroblastoma × glioma hybrid (NG108-15) cells^{17,18} and at human P2Y₂R heterologously expressed in 1321N1 astrocytoma cells.¹⁹ Inhibition of UTP-induced calcium mobilization was determined using the calcium-complexing fluorescent dye Fura-2.^{18,20} Assays were performed using a FLUOstar[®] plate reader (BMG LabTechnologies, Offenburg, Germany) in 96-well plates as previously described.^{18a,21} UTP was used for activating the P2Y₂R at a concentration where it exhibited ca. 50–80% of the maximal effect (1 μM for 1321N1 astrocytoma cells expressing the human P2Y₂R, 3 μM for NG108-15 cells) (for details, see Supporting Information).

The lead structure RB-2 (**1**) exhibited an IC₅₀ value of 5.0 μM at mouse and 1.85 μM at human P2Y₂R in agreement with literature data.⁴ The much smaller, unsubstituted phenylamino-anthraquinone derivative **3** (AB-25) was nearly as potent as RB-2 (11.1 μM at mouse and 5.16 μM at human P2Y₂R). The phenyl residue was essential since the starting compound bromaminic acid (**2**) was inactive. Chloro-substitution in the *meta*-position abolished activity, while a number of other mono- and di-substitutions were tolerated. Potencies ranged from 9 to 30 μM at mouse and 3 to 30 μM at human P2Y₂R for the majority of the compounds. Most anilinoanthraquinones were more potent at human as compared to mouse P2Y₂R. SARs appeared to be slightly different at mouse and at human receptors indicating subtle species-dependent differences of the binding site. The most potent compound of the present series at the human P2Y₂R was the *m*-methyl derivative **5** (IC₅₀ 3.04 μM); however, the compound was 7-fold weaker at mouse P2Y₂R (IC₅₀ 22.0 μM). The *o*-methoxy derivative **11**, which was nearly equipotent at mouse and human P2Y₂R, was selected for further preliminary characterization. It was much weaker in inhibiting rat NTPDase1 (*K*_i > 100 μM) and rat ecto-5'-nucleotidase (*K*_i > 100 μM), and in antagonizing the human P2Y₄ (IC₅₀ ca. 50 μM) and the rat P2Y₆ receptor subtypes

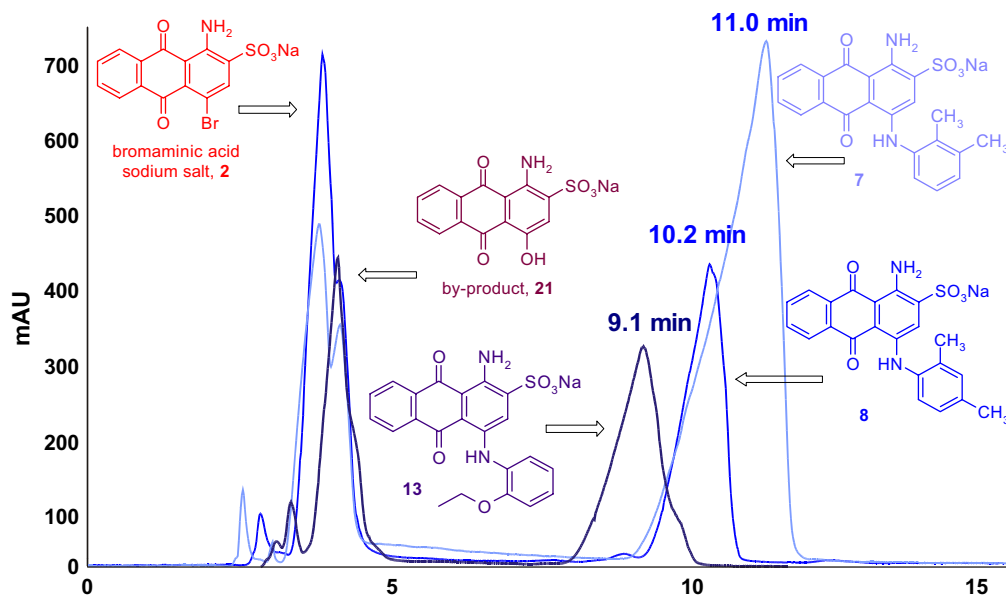


Figure 2. HPLC chromatograms of selected products (**7**, **8**, and **13**, blue). Raw products contain the starting compound (**2**, red-colored) and a by-product (**21**, dark red-colored), which can be removed by preparative HPLC. X-axis: time (min); y-axis: mAU, milli-absorbance units. For conditions, see Ref. 14.

($IC_{50} > 100 \mu M$) (unpublished data) than as P2Y₂ antagonist. However, a comprehensive characterization has yet to be performed.

In conclusion, we established a combinatorial synthetic approach using a compact parallel synthesizer to obtain a small library of 1-amino-4-phenylamino-2-sulfoanthraquinone derivatives related to RB-2 by Ullmann coupling reaction. The method has been used to synthesize three new compounds (compounds **9**, **13**, and **19**) that have, to the best of our knowledge, not been described in the literature. In addition we have synthesized and characterized several compounds (**7**, **8**, **10**, **14**, **17**, and **18**) for which neither synthetic procedures nor spectral data have previously been reported. After having obtained the proof of principle, this reaction can now be transferred to an even larger, fully automated parallel synthesizer to produce extensive compound libraries. Numerous anilines and related amine derivatives are commercially available as building blocks. Furthermore, we have recently developed a synthetic approach to obtain a large variety of substituted *p*-(phenylamino)aniline derivatives²² allowing us to synthesize derivatives of **19**, the compound with the second highest potency of the present series at human P2Y₂R. Very recently we have successfully improved the classical Ullmann coupling reaction by using phosphate buffer, pH 6–7, and performing the reaction under microwave irradiation, which led to dramatically accelerated reactions and improved yields.¹⁵ The next step would be to apply these conditions to parallel combinatorial synthesis. For purification of the products, an efficient preparative HPLC method has been developed. Several small molecules with molecular weight of less than 500 g/mol proved to be P2Y₂R antagonists with low micromolar IC_{50} values. Especially compound **11** (PSB-716) was found to be a promising candidate

since—in contrast to the standard P2Y₂ antagonist Reactive Blue 2 (**1**, RB-2)—it appeared to be selective versus other P2Y subtypes as well as nucleotide-metabolizing enzymes including ecto-5'-NT and E-NTPDase1. Future efforts will be directed toward the replacement of the remaining sulfonate group by carboxylate in order to obtain drugs that will be able to penetrate membranes. Further modification may lead to more potent and selective P2Y₂ antagonists.

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Supplementary data

Supplementary data associated with this article can be found, in the online version, at [doi:10.1016/j.bmcl.2007.10.082](https://doi.org/10.1016/j.bmcl.2007.10.082).

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13. *Synthetic procedure.* Bromaminic acid sodium salt (**2**, 0.250 g, 0.62 mmol), anhydrous copper(II)sulfate (0.015 g, 0.09 mmol), and sodium carbonate (0.085 g, 0.80 mmol) were added to polypropylene reaction vials (MiniBlockTM 48, Mettler Toledo). Then 2.48 mmol of the corresponding aniline derivative (see Table 2) and 2–3 mL of water were added. The reactor was locked and fixed on the shaking-washing unit (Mettler Toledo). The mixtures were shaken for 2 days at 90 °C. Then the vials were filtered and the filter residues were washed 3 times with 2 mL of dichloromethane each. Then the filter residues were suspended in a mixture of water and methanol (50:50), and the water-methanol solution was combined with the aqueous layer, put into round-bottom flasks and the solvent was evaporated under reduced pressure. The residue was taken up in water and washed several times with dichloromethane until the organic phase was clear and colorless. The volume of the aqueous phase was reduced by rotary evaporation and subsequently subjected to lyophilization.
14. *HPLC purification method.* The lyophilized solid products were dissolved in 50 mL of a mixture consisting of water:methanol = 65:35 and filtered through a microfilter (0.45 µm) to remove insoluble materials. The solutions were subsequently subjected to HPLC chromatography using a Knauer HPLC instrument equipped with a pump K1800, a UV detector K-2600, and a Eurospher 100 C 18 column, 10 µm, 250 × 20 mm (ID). An amount of 3 ml containing 30 mg of product was injected and the separation was performed for 15 min at a flow rate of 30 mL/min under the following conditions: isocratic elution (H₂O:methanol = 65:35) for 1 min, followed by a gradient from H₂O:methanol (65:35) to H₂O:methanol (26:74) for 14 min.
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16. Compound **13** was prepared according to the general procedure¹³ using 2-ethoxyaniline (0.339 g, 0.324 mL). Yield: 52%; HPLC retention time: 9.1 min. ¹H NMR (DMSO-*d*₆, 500 MHz, 293 K): δ [ppm] = 1.33 (t, 3H, ³J = 6.9 Hz, CH₃); 4.12 (q, 2H ³J = 7.0 Hz, OCH₂); 6.96–7.02 (m, 1H, C4'H or C5'H); 7.13–7.16 (m, 2H C3'H or C6'H, C4'H or C5'H); 7.28 (d, 1H, ³J = 7.6 Hz, C3'H or C6'H); 7.48 (br, 1H, NH₂); 7.83 (td, 1H, ³J = 7.2 Hz, ⁴J = 2.1 Hz, C6H or C7H); 7.85 (td, 1H, ³J = 7.2 Hz, ⁴J = 2.2 Hz, C6H or C7H); 7.95 (s, 1H, C3H); 8.25–8.29 (m, 2H, C5H, C8H); 10.11 (br, 1H, NH₂); 11.96 (s, 1H, NH). ¹³C NMR (DMSO-*d*₆, 125 MHz, 293 K): δ [ppm] = 14.78 (CH₃); 64.14 (OCH₂); 109.30, 111.57 (C4a, C9a); 113.49 (C3'); 120.95 122.73 (C3); 123.35, 125.19 (C4', C5', C6'); 126.11, 126.15 (C5, C8); 128.42 (C1'); 132.86, 133.18, 133.81, 134.28 (C6, C7, C8a, C10a); 140.74, 142.58, 144.39 (C1, C2, C4); 151.16 (C2'); 181.93, 182.37 (C9, C10). LC-MS m/z (%) = 437.0 (100; [C₂₂H₁₇N₂O₆S][−]); 407.9 (47; [C₂₀H₁₂N₂O₆S][−]); 255.1 (29; [C₁₄H₁₁N₂O₃][−]). Purity by HPLC-UV (254 nm)-ESI-MS: 100%. Anal Calcd for C₂₂H₁₇N₂O₆S·Na·3H₂O: C, 51.36; H, 4.51; N, 5.44. Found: C, 51.39; H, 4.21; N, 5.28. UV absorption (CH₃OH): λ_{max} = 464 nm. Fluorescence (CH₃OH): λ_{max} = 740 nm.
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